

Chemistry 307

Inorganic Chemistry

Spring 2022

Instructor: Dr. Wei-Tsung Lee, Flanner Hall 402A; Ext. 83205; E-mail: wlee5@luc.edu

Lecture: M, W and F 12:35 – 1:25 p.m.; Dumbach Hall 233

Discussion: M 1:40 – 2:30 p.m.; Dumbach Hall 117

Office Hours: W and F, 1:40 – 2:30 p.m. or by appointment

Class Pre-requisite: Chem 222 or 224 and 226

Course Format: This course will be delivered by the in-person format. However, amid a recent surge of the highly transmissible Omicron variant of the coronavirus, the first two weeks (until 1/31/2022) of courses and discussion sections will be delivered through synchronous Zoom platform (password: Chem30722).

Required Textbook: Inorganic Chemistry, 7th Edition, M. Weller, T. Overton, J. Rourke and F. Armstrong, ISBN: 978-0-19-876812-8, Oxford University Press, 2018

Abbreviated solutions to self-tests and exercises from the book can be found online and in Sakai: https://oup-arc.com/access/ichem7e-student-resources#tag_answers-to-self-test-questions

Recommended Materials: Molecular Model Kit, ISBN-09648837-0-8 (2001), by Stephen Darling (see www.molecularvisions.com or www.darlingmodels.com). Model kits for organic chemistry classes do not contain pieces for constructing octahedral and trigonal bipyramidal shapes, etc. which are common in inorganic chemistry. Rather than buying the recommended Molecular Model Kit, some students prefer to access the free website <https://symotter.org/> for visualization of symmetry elements and assignments of point groups.

Course Description and Learning Outcomes: Master basic concepts in inorganic chemistry, such as structure and bonding, transition metal chemistry and organometallics, as well as obtain an appreciation for the role of metal ions in biological systems.

Sakai and Lecture Notes: The instructor will upload lecture notes and slides on Sakai, and will make every effort to have the materials posted on the site at least a day before the lecture. A word of foreknowledge is that the PowerPoint presentations can be quite large (on the order of megabytes) and hence, if you do not have a high-speed internet connection at home, you should consider using Loyola's computer resources to download the materials.

Grading Policy: 100 points for each of the *two 50-min exams*, 25 points for each of the *four 15-min quizzes*, and 200 points for *the final exam* for a grand total of 500 points. The exams will consist of multiple-choice and short-answer questions, but the quizzes will only contain multiple-choice questions. The final exam will be comprehensive with 60% covering material since Exam II and the remaining 40% on the material from Exams I and II. No makeup exams or quizzes will be given. For missed exams, a **written** doctor's or judge's excuse, or a letter from a funeral director, or a notification of a Medical School interview is required; the score for a missed exam or quiz will be determined from the scaled scores in the other exams and quizzes. Exceptions are, however, made for Students involved in Co-Curricular Activities. In those cases, the Loyola University Absence Policy is followed:

Students missing classes while representing Loyola University Chicago in an official capacity (e.g. intercollegiate athletics, debate team, model government organization) shall be allowed by the faculty member of record to make up any assignments and to receive notes or other written information distributed in the missed classes. Students should discuss with faculty the potential consequences of missing lectures and the ways in which they can be remedied. It is the responsibility of the student to make up any assignments. If the student misses an examination, the instructor is required to give the student the opportunity to take the examination at another time. Students must provide their instructors with proper documentation describing the reason for and date of the absence: <https://www.luc.edu/athleteadvising/attendance.shtml>. This documentation must be signed by an appropriate faculty or staff member, and it must be provided as far in advance of the absence as possible.

Class Grades: Total raw scores will also be used to establish final letter grades:

A = 100-85; A- = 84-80; B+ = 79-75; B = 74-70; B- = 69-65; C+ = 64-60; C = 59-55; C- = 54-50; D+ = 49-45; D = 44-40; F = Less than 40

Final Exam: The University sets the schedule for all final exams. The final will be held on 9 am 5/6/2022. You will have exactly 2 hours to complete the exam. Additional time will not be granted, even if you start late. There will be no make-up final exams given under any circumstance, and the exam will not be given early, either. Instructors may not reschedule final exams for a class for another day and/or time during the final exam period. There can be no divergence from the posted schedule of dates for final exams. Individual students who have four (4) final examinations scheduled for the same date may request to have one of those exams rescheduled. If a student reports having four final examinations scheduled for the same date, students should be directed to e-mail a petition to Adam Patricoski, Assistant Dean for Student Academic Affairs, CAS Dean's Office (apatricoski@luc.edu).

Course Repeat Rule: Effective with the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include

withdrawals (W). After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to register form or print it from the Department of Chemistry & Biochemistry website: <http://www.luc.edu/chemistry/forms> and personally meet and obtain a signature from either the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt.

Course/Instructor Evaluation: After the withdrawal deadline (Monday, March 28th) and up to the last day of classes, students will be given the opportunity to evaluate both the instructor and the course by using an online survey. Towards the end of the course, you will receive an email from the Office of Institutional Effectiveness to provide feedback on the course. You will receive consistent reminders throughout the period when the evaluation is open, and the reminders will stop once you have completed the evaluation. The evaluation is completely anonymous. When the results are released, instructors and departments will not be able to tell which student provided the individual feedback. Because it is anonymous and the results are not released to faculty or departments until after grades have been submitted, the feedback will not impact your grade. The feedback is important so that the instructor can gain insight into how to improve teaching and the department can learn how best to shape the curriculum. The essential objective for this course is “Gaining a basic understanding of the subject (*e.g.*, factual knowledge, methods, principles, generalizations, theories)”, and the important objective is “Learning to apply course material (to improve thinking, problem solving, and decisions).

CHEM 307 is an advanced upper-level class and, for Private Tutoring, you may wish to seek the help of an advanced student who has successfully completed the course or of a graduate student who is conducting research in the area of inorganic chemistry.

Student Accommodations: If you have any special needs, please let me know in the first week of classes. The university provides services for students with disabilities. Any student who would like to use any of these university services should contact the Student Accessibility Center (SAC), Sullivan Center, (773) 508-3700. Further information is available at <http://www.luc.edu/sac>

Academic Integrity: All students in this course are expected to have read and to abide by the demanding standard of personal honesty, drafted by the College of Arts & Sciences, which can be viewed at: <http://www.luc.edu/cas/advising/academicintegritystatement>

A basic mission of a university is to search for and to communicate the truth as it is honestly perceived. A genuine learning community cannot exist unless this demanding standard is a fundamental tenet of the intellectual life of the community. Students of Loyola University Chicago are expected to know, to respect, and to practice this standard of personal honesty. Academic dishonesty can take several forms, including, but not limited to cheating, plagiarism, copying another student’s work, and submitting false documents.

Any instance of dishonesty (including those detailed on the website provided above or in this syllabus) will be reported to Professor Miguel Ballicora, the Chairperson of the Department of Chemistry & Biochemistry, who will decide what the next steps may be. Any student found cheating on any examination or quiz will receive a "0" for that assignment. Moreover, depending on the severity of the misconduct, a final grade of F may be assessed for the course. We remind you that such an incident will become part of one's personal record and may be transmitted to organizations, such as medical or dental schools, pharmacy and graduate programs.

Spring 2022 Masking Requirement: It is Departmental policy that, even in the event the University relaxes its universal requirement for indoor mask-wearing during the Spring 2022 semester, it will remain a principle of this class section that, out of respect for the health of housemates and others in regular contact with members of our community, in this class we properly wear masks at all times (*e.g.*, over nose and mouth)

Error Policy: The instructor reserves the right to amend or correct this syllabus.

Schedule and Approximate Syllabus:

Lecture #	Date	Topic	Reading
1Z	1/19	Course Introduction and Atomic Structure	Ch. 1
2Z	1/21	Shielding	Ch. 1
3Z	1/24	Atomic Properties	Ch. 1
4Z	1/26	Molecular Shapes and VSEPR	Ch. 2.1 – 2.3
5Z	1/28	Symmetry Elements	Ch. 3.1
6Z	1/31	Point Groups	Ch. 3.1
7	2/2	Polarity and Chirality	Ch. 3.3, 3.4
8	2/4	VB Theory of Diatomics and Polyatomics	Ch. 2.4 – 2.6
9	2/7	MO Theory of Homo- and Hetero-Diatomics	Ch. 2.7 – 2.9
10	2/9	MO Theory of Polyatomics	See Power Points
11	2/11	Acids and Bases	See Power Points
12	2/14	Nomenclature of Coordination Compounds	Ch. 7.1, 7.2
	2/16	Review	

	2/18	EXAM I (Lectures 1 – 10)	
13	2/21	Coordination Numbers	Ch. 7.3 – 7.6
14	2/23	Isomerism of Coordination Compounds	Ch. 7.7 – 7.10
15	2/25	Crystal Field Theory	Ch. 20.1
16	2/28	Crystal Field Theory (cont.)	Ch. 20.1
17	3/2	Magnetochemistry	Ch. 20.1, 20.8
18	3/4	Crystal Field Theory (cont.)	Ch. 20.1
19	3/14	Ligand Field Theory	Ch. 20.2
20	3/16	Term Symbols	Ch. 20.3
	3/18	Review	
	3/21	EXAM II (Lectures 11 – 19)	
21	3/23	Electronic Spectra	Ch. 20.4 – 20.6
22	3/25	Electronic Spectra (cont.)	Ch. 20.4 – 20.6
23	3/28	Electronic Spectra (cont.)	Ch. 20.4 – 20.6
24	3/30	Substitution Reactions in O_h Complexes	Ch. 21.1-21.2, 21.6 – 21.7
25	4/1	Substitution Reactions in D_{4h} Complexes	Ch. 21.3 – 21.4
26	4/4	Electron Transfer Reactions	Ch. 21.10 – 21.12
27	4/6	Bioinorganic Chemistry	Ch. 26
28	4/8	Bioinorganic Chemistry (cont.)	Ch. 26
29	4/11	Bioinorganic Chemistry (cont.)	Ch. 26
30	4/13	Metals in Medicine	Ch. 27
31	4/20	Metals in Medicine (other)	See Power Points
32	4/22	18- e^- Rule and Organometallic Compounds	Ch. 22.1-22.4
33	4/25	Carbonyl and π -donor Complexes	Ch. 22.5-14, 22.17, 22.18g
34	4/27	Organometallic Rxns & Catalysis	Ch.22.21-26, 22.28, 22.32

4/29

Review

The final examination date: 5/6/2022 Friday, 9 - 11 am (60% on Lectures 20 – 34; 20% on Lectures 1 – 10, and 20% on Lectures 11 – 19).